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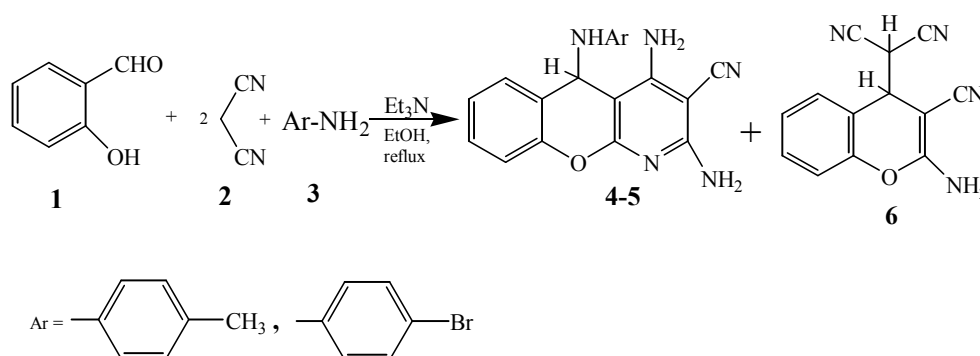
Graphical Abstracts

New synthesis of 2,4-diamino-10-(4-bromo-phenylamino)-10H-9-oxa-1-aza-anthracene-3-carbonitrile, 2,4-diamino-10-p-tolylamino-10H-9-oxa-1-anthracene-3-carbonitrile and 2-(2-amino-3-cyano-4H-chromen-4-yl)-malononitrile

pp 413-416

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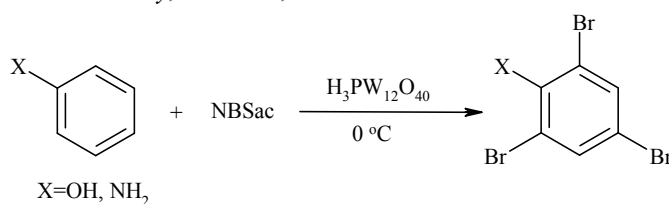


Directed efficient and rapid bromination of phenols and anilines with *N*-bromosaccharin using tungstophosphoric acid as a heterogeneous recyclable catalyst under solvent-free conditions

pp 417-422

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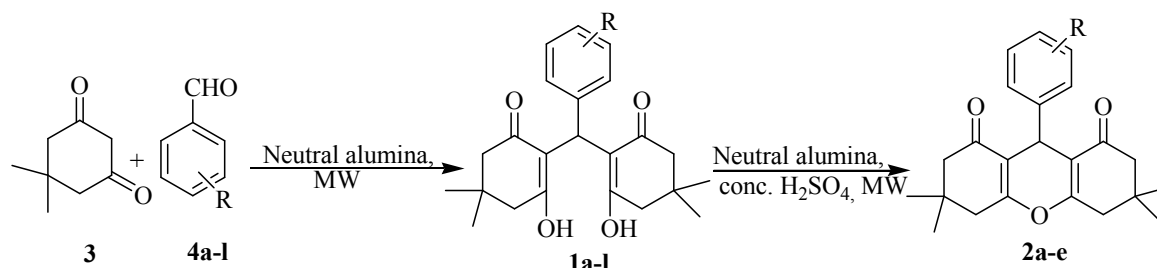


Solvent free solid support synthesis of arylmethylene bis(3-hydroxy-2-cyclohexene-1-ones) and xanthenediones derivatives by microwave irradiation

pp 423-429

Mithu Saha and Amarta Kumar Pal*

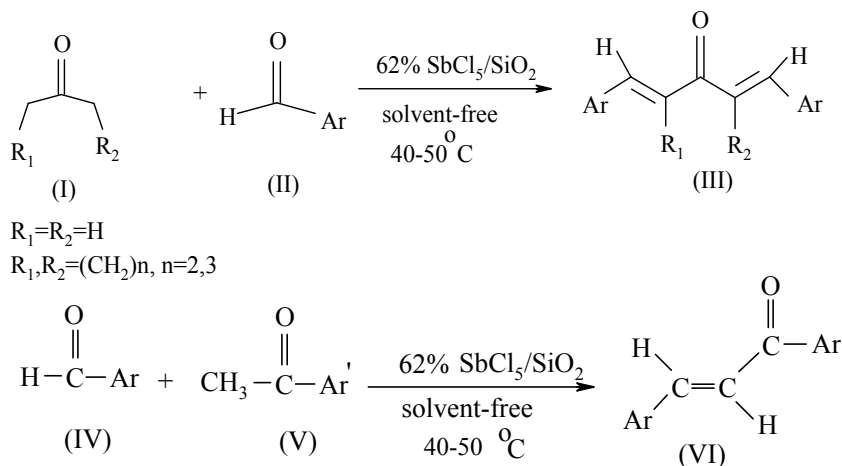
Department of Chemistry, North Eastern Hill University, Shillong 793022, India.



SbCl₅-SiO₂: an efficient reagent system for regio- chemo- and stereoselective claisen- schmidt condensation pp 431-435

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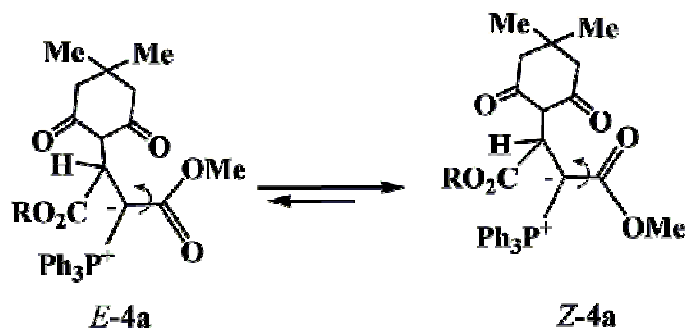


AIM analysis for assignment of the two *Z*- and *E*-isomers in phosphorane containing a dimedon-1-yl pp 437-444

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Identification and removal of amine, cyanide and aldehyde compounds in cigarette smoke via extracted plant's micro silica by gas chromatography – mass spectroscopy (GC/MS) method pp 445-449

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Silica is a compound found in some plants and one of the most important materials in chemical industrial and adsorption processes. In this study, micro silica was extracted from *Equisetum arvense* in North of Iran and purified from *Equisetum arvense* plant ash. SEM measurements determined a particle size of micro silica of 1 –1.5 micrometer. The absorbed compounds were extracted by pure methanol and analyzed by GC/MS, using NIST and WILLY229 libraries for identification.

Electrochemical investigation of *p*-chloranil as an organic mediator in determination of dopamine using of multiwall carbon nanotubes paste electrode pp 451-459

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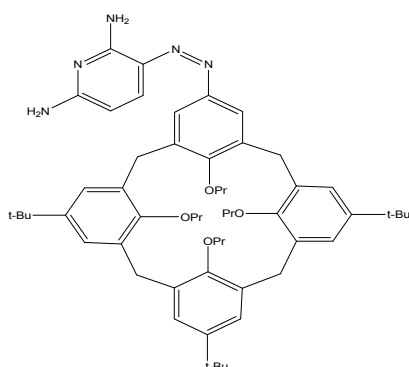
In this work, we study electrochemical behavior of *p*-chloranil as a suitable mediator for voltammetric determination of dopamine (DA) using voltammetric method. Using the modified electrode, the kinetics of DA electrooxidation was considerably enhanced by lowering the anodic over potential through a catalytic fashion. The differential voltammetric peak currents of the electrode increased linearly with the corresponding DA concentration in the range of 0.5-115 $\mu\text{mol L}^{-1}$ with a detection limit of 0.2 $\mu\text{mol L}^{-1}$. The influence of pH and potential interfering substances on the determination of DA were studied. The RSD% for 5 replicate s determination of 100 $\mu\text{mol L}^{-1}$ of DA were 1.5%. Finally, the sensor was examined as a selective, simple, and precise new electrochemical sensor for the determination of DA in real samples, such as drug and urine, with satisfactory results.

Synthesis and characterization of a novel derivative of azo calix [4]-arenes as a new potential for analgesic and antibacterial drug pp 461-464

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Selective Iron (II) colorimetric sensor based on hydroxamic acid-functionalized gold nanoparticles pp 465-472

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