
Research Article**Synthesis of novel [1,2,4]triazolo[4,3-*a*]quinoxaline derivatives**

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nikpour@iauahvaz.ac.ir**ABSTRACT**

Reaction of 2, 3- dichloroquinoxaline with hydrazine in methanol at room temperature gave 1-(2-chloroquinoxalin-3-yl) hydrazine and after 24 hours. Condensation of 1-(2-chloroquinoxalin-3-yl) hydrazine with ethylorthoesters on their boiling carboxylic acids for 8 hours afforded a group of 4-Chloro- [1,2,4] triazolo[4, 3-*a*]quinoxaline derivatives. Stirring of 4-Chloro- [1,2,4]triazolo[4, 3-*a*]quinoxaline derivatives with amines for 2 hours in ethanol at room temperature afforded a group of 4-Amino- [1,2,4]triazolo[4, 3-*a*]quinoxaline derivatives. Solvent effect of methanol and chloroform on this reaction and the spectral data is discussed.

Keywords: 2,3- dichloroquinoxaline; [1,2,4]triazolo[4, 3-*a*]quinoxaline; orthoesters

1. Introduction

Fused quinoxalines are a well-known class of heterocycles. In the recent decades, these class of heterocycles attract attention of chemists for the new biologically active structures [1-5]. Moreover, triazoles and especially fused triazoles are an important class of heterocyclic compounds with varied activities including; antifungal [6], bactericidal [6,7], anxiolytic [8,9], anticonvulsant [10], herbicidal [11], antidepressants [12] and Antiproliferative [13]. Keeping

this in view, we decide to prepare a group of new [1,2,4]triazolo[4,3-*a*]quinoxaline derivatives.

2. Experimental

2.1. Materials and instruments

The melting points were recorded on an Electro thermal type 9100 melting point apparatus. The IR spectra were obtained on a 4300 Shimadzu Spectrometer. The ¹H NMR (300 MHz) spectra were recorded on a Bruker AC 300 spectrometer. The mass spectra were scanned on a Varian Mat CH-7 instrument at 70 eV. Elemental analysis was obtained on a Thermo Finnigan Flash EA micro analyzer. The purity of all new compounds synthesized was tested by TLC using chloroform as mobile phase. Precursors 2 was prepared according to an earlier procedure [14].

2.2. General Procedure for Preparation of 4-Chloro-[1,2,4]triazolo[4,3-*a*]quinoxalines 3a-c.

A mixture of 2-chloro-3-hydrazinyl quinoxaline 2 (0.39 gr, 2 mmol) and either (triethyorthoformate, triethyorthoacetate or triethyorthopropionate, 1.5 ml) in appropriate carboxylic acid (formic acid, acetic acid or propionic acid) (10 ml) heated under reflux condition for 8 hours. The resulting mixture dried by heating and recrystallized from ethanol to achieve novel 4-chloro- [1,2,4] triazolo[4, 3- *a*]quinoxaline derivatives 3a-c.

4-Chloro-[1,2,4]triazolo[4,3-*a*]quinoxaline 3a:

Yield (72%), red powder, mp 152-153 °C. IR spectrum, ν , cm^{-1} : 1570 (C=N). ¹H NMR (d^6 -DMSO) ppm: 8.05 (dd, 2H, C₇H & C₈H); 8.32 (dd, 2H, C₆H & C₉H); 9.8 (s, 1H, C₁H). Mass spectrum, m/z : 204 [M]⁺, 206 [M+2]⁺. Anal. Calcd. for C₉H₅ClN₄: C, 52.83; H, 2.46; N, 27.38 Found: C, 53.05; H, 2.49; N, 27.31.

4-Chloro-1-methyl-[1,2,4]triazolo[4,3-*a*]quinoxaline 3b:

Yield (83%), yellow powder, mp 161-163 °C. IR spectrum, ν , cm^{-1} : 1570 (C=N); 2910, 2940 (CH_3). ^1H NMR (d^6 -DMSO) ppm: 2.37 (s, 3H, CH_3), 8.05 (dd, 2H, C_7H & C_8H); 8.32 (dd, 2H, C_6H & C_9H). Mass spectrum, m/z : 218 $[\text{M}]^+$, 220 $[\text{M}+2]^+$. Anal. Calcd. for $\text{C}_{10}\text{H}_7\text{ClN}_4$: C, 54.93; H, 3.23; N, 25.62 Found: C, 54.82; H, 3.29; N, 25.78.

4-Chloro-1-ethyl-[1,2,4]triazolo[4,3-a]quinoxaline 3c:

Yield (85%), yellow powder, mp 146-148 °C. IR spectrum, ν , cm^{-1} : 1570 (C=N); 2910, 2960 (CH_3). ^1H NMR (d^6 -DMSO) ppm: 0.97 (t, 3H, CH_3), 2.65 (q, 3H, CH_2), 8.05 (dd, 2H, C_7H & C_8H); 8.32 (dd, 2H, C_6H & C_9H). Mass spectrum, m/z : 232 $[\text{M}]^+$, 234 $[\text{M}+2]^+$. Anal. Calcd. for $\text{C}_{11}\text{H}_9\text{ClN}_4$: C, 56.78; H, 3.90; N, 24.08 Found: C, 56.59; H, 3.96; N, 21.18; S, 24.19.

2.3. General Procedure for Preparation of 4-amino-[1,2,4]triazolo[4,3-a]quinoxalines 4a-f.

A mixture of foregoing products 3a-c (1 mmol) and either (morpholine or pyrrolidine 1 ml) in ethanol (5 ml) stirred at room temperature for 2 hours. To the resulting mixture water (10 ml) added after cooling and the precipitant recrystallized from ethanol to achieve novel 4-amino- [1,2,4] triazolo[4, 3- a]quinoxaline derivatives 4a-f.

4-(Morpholin-4-yl)- [1,2,4]triazolo[4,3-a]quinoxaline 4a:

Yield (80%), red powder, mp 171-172 °C. IR spectrum, ν , cm^{-1} : 1570 (C=N) 2910, 2960 (CH_2). ^1H NMR (CDCl_3) ppm: 3.65 (t, 4H, $\text{N}(\text{CH}_2)_2$), 3.91 (t, 4H, $\text{O}(\text{CH}_2)_2$), 7.96 (dd, 2H, C_7H & C_8H); 8.15 (dd, 2H, C_6H & C_9H); 9.7 (s, 1H, C_1H). Mass spectrum, m/z : 255 $[\text{M}]^+$. Anal. Calcd. for $\text{C}_{13}\text{H}_{13}\text{N}_5\text{O}$: C, 61.17; H, 5.13; N, 27.43 Found: C, 61.34; H, 5.18; N, 27.31.

4-(Pyrrolidin-1-yl)-[1,2,4]triazolo[4,3-a]quinoxaline 4b:

Yield (80%), red powder, mp 185-186 °C. IR spectrum, ν , cm^{-1} : 1570 (C=N) 2910, 2960 (CH_2). ^1H NMR (CDCl_3) ppm: 2.05 (t, 4H, $(\text{CH}_2)_2$), 3.62 (t, 4H, $\text{N}(\text{CH}_2)_2$), 7.96 (dd, 2H, C_7H & C_8H); 8.15 (dd, 2H, C_6H & C_9H); 9.7 (s, 1H, C_1H). Mass spectrum, m/z : 239 $[\text{M}]^+$. Anal. Calcd. for $\text{C}_{13}\text{H}_{13}\text{N}_5$: C, 65.25; H, 5.48; N, 29.27 Found: C, 65.14; H, 5.41; N, 29.39.

1-Methyl- 4-(morpholin-4-yl)- [1,2,4]triazolo[4,3-a]quinoxaline 4c:

Yield (70%), red powder, mp 116-117 °C. IR spectrum, ν , cm^{-1} : 1570 (C=N) 2910, 2960 (CH_2). ^1H NMR (CDCl_3) ppm: 2.41 (s, 3H, CH_3), 3.65 (t, 4H, $\text{N}(\text{CH}_2)_2$), 3.91 (t, 4H, $\text{O}(\text{CH}_2)_2$), 7.96 (dd, 2H, C_7H & C_8H); 8.15 (dd, 2H, C_6H & C_9H). Mass spectrum, m/z : 269 $[\text{M}]^+$. Anal. Calcd. for $\text{C}_{14}\text{H}_{15}\text{N}_5\text{O}$: C, 62.44; H, 5.61; N, 26.01 Found: C, 62.32; H, 5.69; N, 25.87.

1-Methyl-4-(pyrrolidin-1-yl)-[1,2,4]triazolo[4,3-a]quinoxaline 4d:

Yield (68%), red powder, mp 124-125 °C. IR spectrum, ν , cm^{-1} : 1570 (C=N) 2910, 2960 (CH_2). ^1H NMR (CDCl_3) ppm: 2.05 (t, 4H, $(\text{CH}_2)_2$), 2.41 (s, 3H, CH_3), 3.62 (t, 4H, $\text{N}(\text{CH}_2)_2$), 7.96 (dd, 2H, C_7H & C_8H); 8.15 (dd, 2H, C_6H & C_9H). Mass spectrum, m/z : 253 $[\text{M}]^+$. Anal. Calcd. for $\text{C}_{14}\text{H}_{15}\text{N}_5$: C, 66.38; H, 5.97; N, 27.65 Found: C, 66.45; H, 6.06; N, 27.52.

1-Ethyl-4-(morpholin-4-yl)-[1,2,4]triazolo[4,3-a]quinoxaline 4e:

Yield (70%), red powder, mp 191-192 °C. IR spectrum, ν , cm^{-1} : 1570 (C=N) 2910, 2960 (CH_2). ^1H NMR (CDCl_3) ppm: 1.02 (t, 3H, CH_3), 2.05 (q, 3H, CH_2), 3.65 (t, 4H, $\text{N}(\text{CH}_2)_2$), 3.91 (t, 4H, $\text{O}(\text{CH}_2)_2$), 7.96 (dd, 2H, C_7H & C_8H); 8.15 (dd, 2H, C_6H & C_9H). Mass spectrum, m/z : 283 $[\text{M}]^+$. Anal. Calcd. for $\text{C}_{15}\text{H}_{17}\text{N}_5\text{O}$: C, 63.59; H, 6.05; N, 24.72 Found: C, 63.68; H, 6.16; N, 24.59.

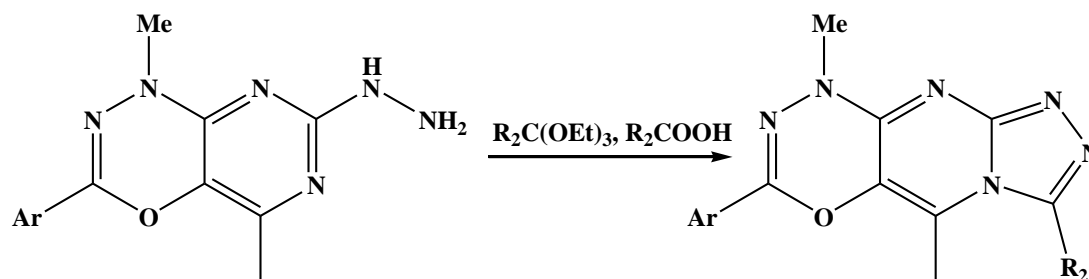
1-Ethyl-4-(pyrrolidin-1-yl)-[1,2,4]triazolo[4,3-a]quinoxaline 4f:

Yield (70%), red powder, mp 185-186 °C. IR spectrum, ν , cm^{-1} : 1570 (C=N) 2910, 2960 (CH_2). ^1H NMR (CDCl_3) ppm: 1.02 (t, 3H, CH_3), 2.05 (t, 4H, $(\text{CH}_2)_2$), 2.05 (q, 3H, CH_2), 3.62 (t, 4H, $\text{N}(\text{CH}_2)_2$), 7.96 (dd, 2H, C_7H & C_8H); 8.15 (dd, 2H, C_6H & C_9H). Mass spectrum, m/z : 267 $[\text{M}]^+$. Anal. Calcd. for $\text{C}_{15}\text{H}_{17}\text{N}_5$: C, 67.39; H, 6.41; N, 26.20 Found: C, 67.54; H, 6.48; N, 26.09.

3. Results and Discussion

In a previous report; treatment of 1-(1,5-dimethyl-3-aryl-1H-pyrimido[4,5-e][1,3,4]oxadiazin-7-yl)hydrazine with orthoesters in their carboxylic acids exhibited as a

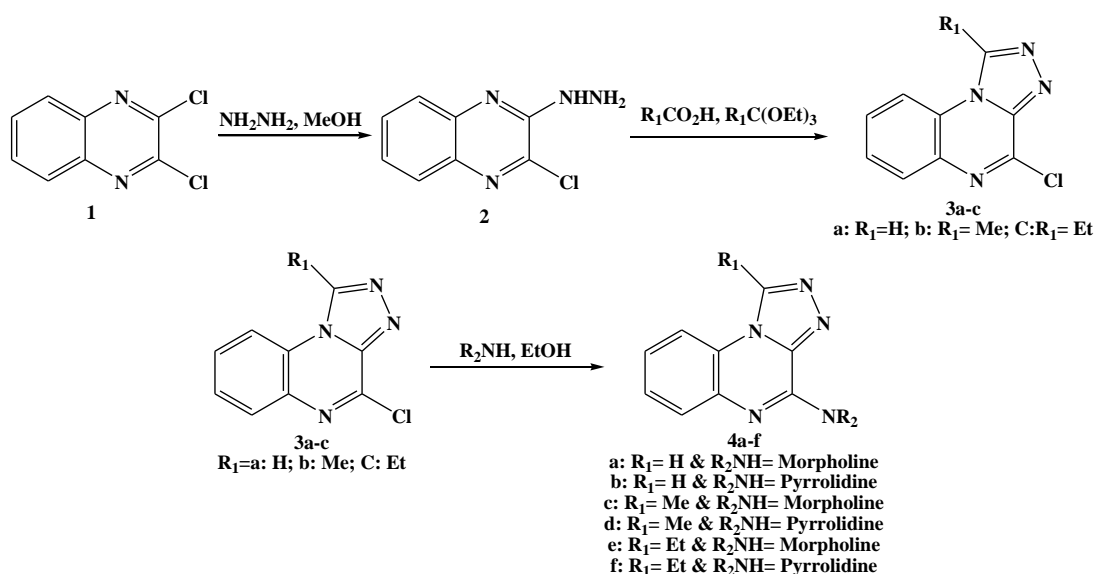
convenient rout for the synthesis of 3-Aryl-1,5-dimethyl-1H-[1,2,4]triazolo[4',3':1,2]pyrimido[4,5-*e*][1,3,4]oxadiazines derivatives [13] as shown in Scheme 1.



Scheme 1

Scheme 1. synthesis of 3-Aryl-1,5-dimethyl-1H- [1,2,4]triazolo[4',3':1,2]pyrimido[4,5-*e*][1,3,4]oxadiazines derivatives

In this study, we extend the foregoing rout for the synthesis of our target compounds. For this purpose; 2,3-dichloroquinoxaline 1 reacted with hydrazine to produce 2-chloro-3-hydrazinyl quinoxaline 2 in ethanol at room temperature according to an earlier procedure [14], which condensed with ethyorthoesters in their boiling carboxylic acids to achieve 4-chloro- [1,2,4] triazolo[4, 3- *a*]quinoxaline derivatives 3a-c. The latter compounds easily reacted with amines in room temperature to produce appropriate 4-amino derivatives 4a-f as shown in *Scheme 2*.



Scheme 2

Scheme 2. synthesis route for achieve 4-chloro- [1,2,4] triazolo[4, 3- *a*]quinoxaline derivatives

The structure of novel derivatives 3a-c and 4a-f were strongly confirmed by their spectral and micro analytical data. The IR spectra did not show the stretching vibration bands at 3450 and 3300 cm^{-1} (broad, NH_2) belonging to precursor 2 but appeared new bands around 2900 cm^{-1} belonging to CH_2 & CH_3 groups. Mass spectra of compounds 4a-f devoid the isotopic effect of chlorine atom of precursor's 3a-c in the molecular ion region and confirm its replacement by amines. ^1H NMR spectra of products 3a-c lacked the broad NH_2 and NH signals at δ 4.2 & 6.5 ppm respectively belonging to precursor 2 but showed new signals at δ 9.8 ppm for 3a and δ 1-2.7 ppm for 3b-c. ^1H NMR spectra of products 4a-f showed new multiple signals due to amino moieties. Microanalytical data of all products showed no significant difference with the calculated values. These results strongly confirms the formation of [1,2,4] triazolo[4, 3- *a*]quinoxaline derivatives.

4. Conclusion

In conclusion, sequential treatment of 1-(4-chloro-3-quinoxalyl)hydrazine with orthoesters in boiling carboxylic acids and amines in ethanol at room temperature is a new, convenient and general access to [1,2,4] triazolo[4, 3- *a*]quinoxaline derivatives.

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References:

- [1] Y. A. Ammar, I. M. Ismail, A. M. El- Sharief, Y. A. Mohamed and R. M. Amer, *J. Ind. Chem. Soc.* 66(1989) 124.
- [2] V. G. Baklykov, V. N. Charushin, O. N. Chupakhin and V. N. Droazd; *Khim. Getrosikl. Soedin.* 4(1987) 557-561.
- [3] D. S. Yufit, Yu. T. Strachkov, V. N. Droazd, V. N. Charushin, V. G. Baklykov and O. N. Chupakhin, *Khim. Getrosikl. Soedin.* 5(1987) 701-706.
- [4] A. J. Elliot and M. S. Gibson; *J. Org. Chem.* 45(1980) 3677-3681.
- [5] M. Bakavoli, M. Nikpour and M. Rahimizadeh; *Phosphorus, Sulfur Silicon Relat. Elem.* 180 (2005) 2265-2268.
- [6] S. A.El-Hawash, N. S.Habib, N. H. Fanaki, *Pharmazie.* 54(1999) 808.
- [7] D. J.Brown, Y. Iwai, *Aust J Chem*,32 (1997) 2727.
- [8] G.Tarzia, E.Ocelli, E.Toja, D.Barone, N.Corsico, L.Gallico, F.Luzzani, *J Med Chem*, 31 (1988) 1115.
- [9] R. I.Trust, J. D. Albright, *U.S. Pat.*4,242,515, (1980) (R. I.Trust, J. D. Albright, *Chem Abstr*, 94 (1981)139815d).
- [10] G.Tarzia, E.Ocelli, D.Barone, *Farmaco*, 44 (1989) 3.
- [11] (a) R.Peignier, A.Chene, R.Cantegril, J. Mortier, *Eur. Pat.* 441718 ,1991,; *Chem Abstr*, 115 (1991) 208000; (b) R.Cantegril, A.Chene, J.Mortier, R.Peignier, *Eur. Pat.* 483027,1992,; *Chem Abstr*, 117(1992)131214.
- [12] S.Sarges, H. R.Howard, R. G.Browne, L. A.Lebel, P. A.Seymour, B. K. Koe, *J Med Chem*, 33(1990) 2240.

- [13] M.Bakavoli, M.Rahimizadeh, A,Shiri, M.Akbarzadeh, S. H.Mousavi, Z. Tayarani-Najaran, H. Atapour-Mashhad, M.Nikpour, *J Heterocycl Chem*, 48(2011) 183.
- [14] R. Sarges, H. R. Howard, R. G. Browne, L. A. Lebel, P. A. Seymour, B. K. Koe; *J. Med. Chem.* 33(1990) 2240.