Gravimetric storage capacity of Hydrogen on $C_{24}H_{12}$ Coronene and its Si substituted at 298 K, a Monte Carlo Simulation

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ABSTRACT: In this study, the radial distribution and gravimetric storage capacities of hydrogen on coronene ($C_{24}H_{12}$) and its Si substituted forms, $C_{24}H_{12}$, $C_{24n}S$ inH₁₂ (n= 4-24), have been investigated at (298 K and 0.1 MPa (standard situation) using (N.V.T) Monte Carlo simulation by Lennard-Jones (LJ) 12-6 potential. The results show that the increase of number of silicon substitution doesn't have any effect on the value of Radial Distribution Function (RDF) of ${\sf H}_{_2}$. Also, the symmetric silicon substituted structures have more RDF than asymmetric types. The magnitude of hydrogen radial distribution on the surface of $\mathsf{C}_6\mathsf{Si}_{\cdot 8}\mathsf{H}_{\cdot 12}$ is larger than other structures. The maximum value of radial distribution, gravimetric storage capacity and weight percent of hydrogen storage under identical temperature and pressure conditions are 0.0434, 0.0612 and %6.12 for $\mathsf{C_6Si_{18}H_{12}}$ structure at 0.73Å. On the basis of results, the coronene with 18Si can be suggested as a choice for hydrogen container compared with other its Si substituted forms.

Symmetry ,distribution Radial ,Hydrogen ,Capacity Storage Gravimetric ,Silicon ,Coronene **:Keywords**

INTRODUCTION

The use of hydrogen as fuel for cars in the future is discussed. Pure hydrogen may be the final destination in the evolution of fuel usage from coal to petroleum ing hydrogen content (Zhou, 2005). A major bottleneck to natural gas, which has followed a trail of increasfor the hydrogen vehicle is the problem of hydrogen storage. Hydrogen has the highest energy per unit mass but occupies a large volume. Even in liquid form, the energy density of hydrogen is only 8.4 MJ/L, compared nized as an ideal energy carrier with heating value three to 34.2 MJ/L for gasoline. Hydrogen has been recogtimes higher than petroleum (Moriatry and Honnery, quire a rather complex combination of materials that 2009). The energy conversion and storage systems reficiencies close to the thermodynamic limit (Bisquert, can selectively promote an adsorption process with ef-2011). Recently, considerable attention has been driven to porous materials such as clathrates, zeolites, carbon als for hydrogen storage (Jena, 2011). Coronene, $C_{24}H_{12}$ nanotubes, fullerenes and coronene as possible materi-(or superbenzene) is a yellow polycyclic aromatic hy-
drocarbon comprising six peri-fused benzene rings that dissolves in such solvents as benzene, toluene, and di-

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chloromethane, Fig. 1-a (Fetzer, 2000, Micha, et al., phitic nanoribbons were synthesized by Talyzin et al 2006). Corenene and a series of coronene-derived graat 2012 (Talyzin, et al., 2011, Fujihara, et al., 2012, De Aguiar, et al., 2014) and enthalpies of formation for these compounds was studied by Thomas (Allison, et al., 2015, Thomas, 2015). The properties of silicon substituted of nano carbon structures have motivated a number of studies in recent years on SiC structures *(Melinon, et al., 2007, Ray, et al., 2006).*

SIMULATION DETAILS

In this work, the radial distribution of hydrogen on coronene $(C_{24}H_{12})$ and its Si substituted forms, $C_{24}H_{12}$, C_{24-n} Si_nH₁₂ (n= 4-24), Fig. 1, have been investigated at ulation by Lennard-Jones (LJ) 12-6 potential $(\phi_{\text{L}_\text{U}_\text{U}_\text{U}})$, 298 K and 0.1 MPa using (N, V, T) Monte Carlo simEq.1 (13 Cheng, *et al.*, 2004).

$$
\phi_{LJ(r)} = 4\varepsilon_{gc} \left[\left(\frac{\sigma_{gc}}{r} \right)^{12} - \left(\frac{\sigma_{gc}}{r} \right)^{6} \right]
$$
 (1)

Where r is the distance between gas molecules and surface atoms in the coronene structures, σ_{sc} denotes the LJ gas- coronene collision diameter, and ε_{gc} is the Jones parameters for the interaction between the gas LJ gas-carbon potential well depth. The Lennardand coronene surfaces were calculated using the Eq. 2, 3 (Lithoxoos, et al., 2008).

$$
\varepsilon_{ij} = \sqrt{\varepsilon_i \varepsilon_j} \tag{2}
$$

$$
\sigma_{ij} = \frac{(\sigma_i + \sigma_j)}{2} \tag{3}
$$

We considered these parameters for Si-C, C-C, H_2 - H_2 , C-H₂ and Si-H₂ pairs (Darkrim, et al., 1998, Lee, et

Fig. 1. Coronene (C $_{\rm 24}$ H $_{\rm 12}$) and its Si substituted C $_{\rm 24n}$ Si $_{\rm n}$ H $_{\rm 12}$ (n= 4-24) forms

al., 2000, Levesque, *et al.*, 2002, *Dimitrakakis, <i>et al.*, tions are imposed in all directions. A cubic simulation 2008). In our simulations, periodic boundary condibox $(50.0 \text{ Å}, 50.0 \text{ Å}, 50.0 \text{ Å})$ contains one coronene uating the LJ potential is set to 2.5 Å on all coronene structure and H2 molecules. The cutoff length in evalstructures. For every calculation, $10⁷$ configurations are generated. The initial configurations are discarded tation and comparison of gravimetric storage capacity semble properties. The aim of present work is presenfigurations are used as the average of the desired ento guarantee equilibration, whereas the remaining con-(absolute value adsorption per mass of adsorbent, ρw) and radial distribution function (RDF) parameters for

Fig. 2. RDF of $\mathsf{C}_{\mathsf{24}}\mathsf{H}_{\mathsf{12}}$, $\mathsf{C}_{\mathsf{6}}\mathsf{Si}_{\mathsf{18}}\mathsf{H}_{\mathsf{12}}$ (isomer A) and $\mathsf{Si}_{\mathsf{24}}\mathsf{H}_{\mathsf{12}}$

the structures. The gravimetric storage capacities on the basis of hydrogen absorption are comparable for all structures. The gravimetric storage capacity, pw, was calculated by Eq. 4.

$$
\rho_{\rm w} = \frac{N_{\rm gas} m_{\rm gas}}{N_{\rm gas} + N_{\rm e} \cdot m_{\rm e} + N_{\rm hetero} \cdot m_{\rm hetero}} \tag{4}
$$

Where N_{gas} , N_c and N_{hetro} are the number of gas molecules outside the coronene structure up to 2.5 Å, numcules outside the coronene structure up to 2.5 Å , number of carbon atoms in the simulation box and number of heteroatom (Si), and m_{gas} , m_c and m_{hetero} (g. mol⁻¹) are the corresponding molar mass, respectively (Ra-
fati, *et al.*, 2010).

RESULTS AND DISCUSSION

The radial distribution of H₂ on C₂₄H₁₂ and C_{24-n}Si_nH₁₂ $(n= 4-24)$ coronene structures at 298 K and 0.1 MPa lecular simulation and fitted by the Lennard-Jones potential equation. of pressure are studied by (N, V, T) Monte Carlo Mo-
lecular simulation and fitted by the Lennard-Jones poof pressure are studied by (N,V,T) Monte Carlo Mo*Gravimetric storage capacity of Hydrogen on C₂₄H₁₂ Coronene ...*

Fig. 3. The values of RDF (max) of structures

stituted forms of corenene with different number and In the present work we have considered the Si subdifferent position of Si atoms. We have investigated the effect of number, position of Si substitution and symmetry of structure on the radial distribution of H2. Therefore, the structures of $C_{24n}Si_nH_{12}$ (n= 6, 8, 9, 10, metry forms. Because of numerousness of structures, 12, 18) have been considered at symmetry and asym-RDF of only three samples, $C_{24}H_{12}$, $C_6Si_{18}H_{12}$ (iso mer A) and $\text{Si}_{24}\text{H}_{12}$ with respect to r (Å) at 298 K are shown at Fig. 2. The maximum of RDF for structures

are shown at Fig. 3. Comparison of RDF (max) of iso-
mer structures of $C_{24n}Si_nH_{12}$ (n= 6, 8, 9, 10, 12, 18) at are shown at Fig. 3. Comparison of RDF (max) of iso-Fig. 3 and Table 1, show that the symmetric structures have more RDF (max) value with respect to asymmet-
ric-type. As an example, $C_{18}Si_{\phi}H_{12}$ (isomer A), (Fig. have more RDF (max) value with respect to asymmet-1-c), with RDF (max) equal to 0.0344 is more symmetric than $C_{18}Si_{\phi}H_{12}$ (isomer B) with 0.0284 of RDF 1-c), with RDF (max) equal to 0.0344 is more sym-(Fig. 1-d). The RDF (max) of $C_{14}Si_{10}H_{12}$ (isomer D) (0.0323), $C_{12}Si_{12}H_{12}$ (isomer D) (0.0323) and $C_9Si_{15}H_{12}$ (0.0328) are almost equal to RDF (max) of $C_{24}H_{12}$ corenene (0.0324). The RDF (max) of $C_{18}Si_{6}H_{12}$ (iso-

Table 1: RDF (max), r (distance between gas and surface of corenene at RDF (max), Gravimetric storage capacity (ρw) and weight percent of hydrogen storage for structures at 298 K and 0.1 MPa

Structure	RDF(max)	r(A)	ρ_w	$wt\%$
$C_{24}H_{12}$	0.0324	0.79	0.0490	4.90
$C_{20}Si_4H_{12}$	0.0310	0.73	0.0481	4.81
$C_{18}Si6H12$ (isomer A)	0.0344	0.75	0.0395	3.95
$C_{18}Si6H12$ (isomer B)	0.0284	0.73	0.0331	3.31
$C_{16}Si_8 H_{12}$ (isomer A)	0.0270	0.71	0.0452	4.52
$C_{16}Si_8H_{12}$ (isomer B)	0.0312	0.73	0.0593	5.93
$C_{16}Si_8H_{12}$ (isomer D)	0.0300	0.73	0.0512	5.12
$C_{15}Si_0H_{12}$ (isomer A)	0.0306	0.71	0.0402	4.02
$C_{15}Si_0H_{12}$ (isomer B)	0.0300	0.73	0.0363	3.63
$C_{14}Si_{10}H_{12}$ (isomer A)	0.0261	0.73	0.0397	3.97
$C_{14}Si_{10}H_{12}$ (isomer B)	0.0301	0.75	0.0463	4.63
$C_{14}Si_{10}H_{12}$ (isomer D)	0.0323	0.73	0.0484	4.84
C_1, Si_1, H_1 (isomer A)	0.0289	0.73	0.0305	3.05
C_1, Si_1, H_1 (isomer B)	0.0339	0.73	0.0537	5.37
$C_1, Si_1, H_1,$ (isomer D)	0.0323	0.73	0.0550	5.55
$C_{10}Si_{14}H_{12}$	0.0306	0.73	0.0542	5.42
$C_0Si_{15}H_{12}$	0.0328	0.73	0.0586	5.86
$C_8Si_{16}H_{12}$	0.0295	0.73	0.0420	4.20
$C_6Si_{18}H_{12}$ (isomer A)	0.0434	0.73	0.0612	6.12
$C_6Si_{18}H_{12}$ (isomer B)	0.0309	0.75	0.0599	5.99
$Si_{24}H_{12}$	0.0290	0.90	0.0502	5.02

mer A) (0.0344), C_1, Si_2H_1 (isomer B) (0.0339), and $C_6Si_{18}H_{12}$ (isomer A) (0.0434) are more than $C_{24}H_{12}$ corenene (0.0324) .

The structures of $C_{20}Si_4H_{12}$, $C_{18}Si_6H_{12}$ (isomer B), $C_{16}Si_8H_{12}$ (isomers A, B, D), $C_{15}Si_9H_{12}$ (isomers A, B), $C_{14}Si_{10}H_{12}$ (isomers A, B), $C_{12}Si_{12}H_{12}$ (isomer A), $C_{10}Si_{14}H_{12}$, $C_8Si_{16}H_{12}$, $C_6Si_{18}H_{12}$ (isomer B) and $Si_{24}H_{12}$ have less RDF (max) than $C_{24}H_{12}$. Fig. 3 shows that the maximum RDF for $C_6Si_{18}H_{12}$ (isomer A) is more than the others. The maximum value of radial distribution under identical temperature and pressure conditions is 0.0434 for $C_6Si_18H_12$ (isomer A) (Fig. 1-u) structure at 0.73Å. Table 1 shows that the calculated form of gravimetric storage capacity (ρw) for $C_6Si_{18}H_{12}$ (isomer A) at 298 K and 0.1 MPa of pressure is bigger age capacity and weight percent of hydrogen storagethan other structures. The values of gravimetric storare obtained for $C_6Si_{18}H_{12}$ (isomer A), are 0.0612 and $%6.12$, respectively. The U.S. Department of Energy (DOE) is proposed the 6.5 $wt\%$ H2 / kg system (wt%) is weight percent) for hydrogen storage at $2005-2010$ (Bouza, et al., 2004).

CONCLUSIONS

In this work, the RDF, ρ_w and %wt of hydrogen on $C_{24}H_{12}$ coronene and its silicon substituted structures have been investigated. The main aim of this study is con substitution on radial distribution function (RDF) to investigate the effect of number and position of siliand gravimetric storage capacity (ρ) of hydrogen. ed structures have more RDF than asymmetric types. The results show that the symmetric silicon substitut-The increase of number of silicon substitution doesn't have any effect on the value of RDF. The maximum value of radial distribution, ρw and %wt are 0.0434, 0.0612 and %6.12 for $C_6Si_{18}H_{12}$ (isomer A) (Fig. 1-u) structure at 0.73Å.

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