

Green synthesis of thiazolo oxazin using multicomponent reactions of thiazole

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Abstract: In this research work, the reaction of activated acetylenic compounds, alkyl bromids and thiazole as nucleophiles was described that was led to synthesis of thiazolooxazine derivatives. Some advantages of this procedure are performing reactions in green media, easy separation of product and high yields of product.

Keywords: Thiazole; Phenacyl bromids; Activated acetylenic compounds, Thiazolooxazin.

Introduction

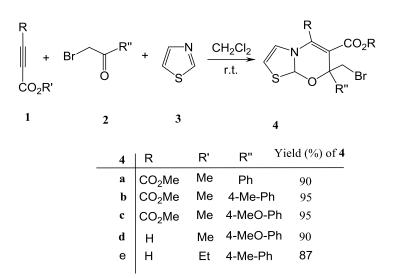
In this respect, several studies involved sulfurcontaining compounds because they present good conduction in organic materials [10,11] or are relevant biologically. Also, sulfurcontaining anions have found extensive use as versatile reagents in organic synthesis. Some heterocyclic compounds containing a thiazole ring in their structures offer important applications in pharmaceutical as well as in agrochemical chemistry [12,13]. For example, ritonavir, an anti-HIV drug contains the thiazole moiety. These products, which have N and S atoms, are bridged easily with other molecules [14,15] or can coordinate several metal ions. For example, they could be used to entrap mercury in the environment [16] and as a new inhibitor for copper [17]. In general, multi-component reactions (MCRs) economically environmentally are and verv advantageous because multi-step syntheses produce considerable amounts of waste mainly due to complex isolation procedures often involving expensive, toxic and hazardous solvents after each step.

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MCRs are perfectly suited for combinatorial library syntheses, thus are finding increasing use in the discovery process for new drugs and agrochemicals [1-7]. In recent years, the research into novel active organic substances and into the design of molecular electronic devices has attracted considerable interest [8,9]. Herein, we describe an efficient procedure for direct synthesis of thiazolooxazin using the three component reaction of activated acetylenic compounds with alkyl bromids in the Presence of thiazole in dichloromethane at room temperature (Scheme 1).

Results and discussion

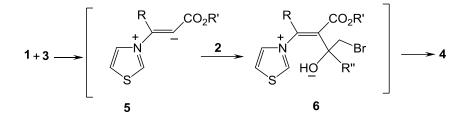
The reaction of activated acetylenic compounds 1 with thiazole 3 in the presence of alkyl bromids 2 led to thiazolooxazin 4 in 87-95% yields (Scheme 1).



Scheme 1: synthesis of thiazole derivatives

Structures of compounds **4a–d** were assigned by IR, ¹H NMR, ¹³C NMR and mass spectral data. For example, the ¹H NMR spectrum of **4a** exhibited one triplet at 1.25 (${}^{3}J_{HH} = 7.2$) for methyl proton and two singlets at 3.68 and 3.89 for methoxy groups. Because of stereogenic center in these products, hydrogens of CH₂ and OCH₂ groups are diasterotopic, therefore, two doublets were observed at 4.09 (${}^{2}J_{HH} = 10.9$) and 4.17 (${}^{2}J_{HH} = 10.9$) for CH₂ group, one multiplet at 4.18-4.25 for OCH₂ moiety and one singlet at 6.60 ppm for CH groups. The carbonyl groups resonances in the ¹³C NMR spectra of **4a** appear at 162.9, 164.1 and 167.5 ppm. The mass spectrum of **4a** displayed the molecular ion peak at m/z = 422.

A tentative mechanism for this transformation is proposed in Scheme 2. It is conceivable that, the initial event is the formation of the 1:1 adducts 5 from the Reaction of activated dialkyl acetylenedicarboxylates 1 with thiazol 3 which is subsequently attacked by alkyl bromids 2 to produce 6. Intermediate 6 undergoes cyclization reaction to generate 4.



Scheme 2: Proposed mechanism for synthesis of 4

Conclusion

In conclusion, the reaction of deficient acetylenic compounds with pyruvates in the presence of 1-methyl thiazole led to thiazolo oxazin in excellent yields. The present procedure has the advantage that the reaction is performed under neutral conditions, and the starting material can be used without any activation or modification.

Experimental

All compounds in these reactions were obtained from Fluka and were used without further purification. M.p.: Electrothermal-9100 apparatus; uncorrected. IR Spectra: Shimadzu IR-460 spectrometer. ¹H and ¹³C NMR spectra: Bruker DRX-500 AVANCE instrument; in CDCl₃ at 500.1 and 125.7 MHz, respectively; \Box in ppm, *J* in Hz. EI-MS (70 eV): Finnigan-MAT-8430 mass spectrometer, in m/z. Elemental analyses (C, H, N) were performed with a Heraeus CHN-O-Rapid analyzer.

General Procedure for the Preparation of Compounds 4a-d

1-Methyl thiazole (2 mmol) were added to a mixture of pyruvates (2 mmol) and activated acetylenic ester (2 mmol) at room temperature. The reaction mixture was then stirred for 2 h to afford the pure compounds **4a-d**.

7-ethyl 5,6-dimethyl 7-bromomethyl-7H-[1,3]thiazolo[2,3-b][1,3]oxazin-5,6,7-tricarboxylate (4a):

Yellow oil, yield: 0.76 g (90%). IR (KBr): 1725, 1591, 1549, 1473, 1368, and 1015. ¹H NMR: 1.25 (3 H, *t*, ${}^{3}J_{HH} = 7.2$, Me), 3.68 (3 H, s, OMe), 3.89 (3 H, s, OMe), 4.09 (1 H, *d*, ${}^{2}J_{HH} = 10.9$, CH), 4.17 (1 H, *d*, ${}^{2}J_{HH} = 10.9$, CH), 4.18-4.25 (2 H, *m*, OCH₂), 5.69 (1 H, *d*, ${}^{3}J_{HH} = 4.5$, CH), 6.19 (1H, *d*, ${}^{3}J_{HH} = 4.5$, CH), 6.60 (1 H, *s*, CH). ¹³C NMR: 13.9 (Me), 31.8 (CH₂Br), 51.9 (OMe), 52.1 (OMe), 62.9 (OCH₂), 79.6 (C), 91.0 (CH), 102.7 (CH), 109.3 (C), 128.8 (CH), 141.7 (C), 162.9 (C=O), 164.1 (C=O), 167.5 (C=O). EI-MS: 422 (M⁺, 10); 350 (20), 348 (20), 167 (25), 149 (60), 84 (100), 57 (62). Anal. Calcd for C₁₄H₁₆BrNO₇S (422.24): C, 39.82; H, 3.82; N, 3.32; found: C, 39.80; H, 3.80; N, 3.31%.

7-ethyl 5,6-diethyl 7-bromomethyl-7H-[1,3]thiazolo [2,3-b][1,3]oxazin-5,6,7-tricarboxylate (4b):

Yellow Oil, yield: 0.76 g (85%). IR (KBr): 1732, 1685, 1583, 1504, 1453 and 1384. ¹H NMR: 1.22 (3 H, t, ${}^{3}J_{HH} = 7.2$, Me), 1.28 (3 H, t, ${}^{3}J_{HH} = 7.2$, Me), 1.35 (3 H, t, ${}^{3}J_{HH} = 7.2$, Me), 4.12 (1 H, d, ${}^{2}J_{HH} = 10.5$, CH), 4.18 (1 H, d, ${}^{2}J_{HH} = 10.5$, CH), 4.19-4.23 (4 H, m, 2 OCH₂), 4.29-4.37 (2 H, m, OCH₂), 5.71 (1H, d, ${}^{3}J_{HH} = 4.6$, CH), 6.20 (1H, d, ${}^{3}J_{HH} = 4.6$, CH), 6.62 (1 H, s, CH). ¹³C NMR: 13.8 (Me), 13.9 (Me), 14.0 (Me), 35.7 (CH₂Br), 61.0 (OCH₂), 62.4 (OCH₂), 62.7 (OCH₂), 78.4 (C), 90.9 (CH), 102.5 (CH), 113.4 (C), 121.4 (CH), 142.0 (C), 162.4 (C=O), 163.6 (C=O), 167.6 (C=O). EI-MS: 450 (M⁺, 5); 377 (24), 375 (24), 370 (68), 231 (45), 229 (45), 84 (100), 73 (60). Anal. Calcd for C₁₆H₂₀BrNO₇S (450.30): C, 42.68; H, 4.48; N, 3.11; found: C, 42.70; H, 4.50; N, 3.10%.

7-ethyl 5,6-dimethyl 7-methyl-7H-[1,3]thiazolo[2,3b][1,3]oxazin-5,6,7-tricarboxylate (4c):

Yellow Oil, yield: 0.58 g (85%). IR (KBr): 1716, 1687, 1429, 1364, 1199 and 1103. ¹H NMR: 1.17 (3

H, t, ${}^{3}J_{\text{HH}} = 7.2$, Me), 1.75 (3 H, s, Me), 3.65 (3 H, s, OCH₃), 3.82 (3 H, s, OCH₃), 4.12-4.17 (2 H, m, OCH₂), 5.61 (1 H, d, ${}^{3}J_{\text{HH}} = 4.6$, CH), 6.11 (1 H, d, ${}^{3}J_{\text{HH}} = 4.6$, CH), 6.52 (1 H, s, CH). 13 C NMR: 13.6 (Me), 23.6 (Me), 51.7 (OCH₃), 53.0 (OCH₃), 61.9 (OCH₂), 89.9 (C), 90.7 (CH), 101.3 (CH), 112.7 (C), 121.4 (CH), 138.4 (C), 163.1 (C=O), 164.5 (C=O), 169.8 (C=O). EI-MS: 343 (M⁺, 10); 270 (85); 306 (66); 292(64), 284 (60);275 (85), 84 (100); 59 (67). Anal. Calcd for C₁₄H₁₇NO₇S (343.35): C, 48.97; H, 4.99; N, 4.08; found: C, 48.95; H, 4.92; N, 4.02%.

7-ethyl 5,6-diethyl 7-methyl-7H-[1,3]thiazolo[2,3b][1,3]oxazin-5,6,7-tricarboxylate (4d):

Yellow Oil, yield: 0.59 g (80%). IR (KBr): 1716, 1686, 1461, 1360, 1312 and 1025. ¹H NMR: 1.16 (3 H, *t*, ³*J*_{HH} = 7.2, Me), 1.19 (3 H, *t*, ³*J*_{HH} = 7.2, Me), 1.27 (3 H, *t*, ³*J*_{HH} = 7.2, Me), 1.71 (3 H, s, Me), 4.00-4.18 (4 H, *m*, 2 OCH₂), 4.20-4.32 (2 H, *m*, OCH₂), 5.58 (1 H, *d*, ³*J*_{HH} = 4.6, CH), 6.07 (1 H, *d*, ³*J*_{HH} = 4.6, CH), 6.52 (1 H, *s*, CH). ¹³C NMR: 13.8 (Me), 13.9 (Me), 14.0 (Me), 23.8 (Me), 60.8 (OCH₂), 61.7 (OCH₂), 62.5 (OCH₂), 78.2 (C), 90.7 (CH), 101.3 (CH), 112.5 (C), 121.6 (CH), 138.8 (C), 162.8 (C=O), 163.9 (C=O), 170.0 (C=O). EI-MS: 371 (M⁺, 15); 298 (85); 225 (66); 292(64), 275 (85), 84 (100); 45 (84). Anal. Calcd for C₁₆H₂₁NO₇S (371.41): C, 51.74; H, 5.70; N, 3.77; found: C, 51.70; H, 5.68; N, 3.71%.

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