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Germination, Growth and Uptake of Heavy Metals in Contaminated Soils (*Hordeum bulbosum* L.)

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Abstract. The use of metal-accumulating plants to clean the contaminated soil with toxic metals is the most rapidly developing component of phytoremediation as an environmentally friendly technology. Overcoming harsh soil conditions and accelerating the recovery of degraded soils remain a worldwide restoration challenge. This study evaluated the effect of different concentrations of Pb, Zn, Cu and Fe on the germination, growth and heavy metal uptake of *Hordeum bulbosum* L. Results obtained from the current research indicated that the plant gives dose-dependent responses to the contaminated soils. Reduction in germination, root and shoot height and biomass were significantly (P<0.05) different as compared to the control. Exposures of heavy metals in the treatment reduced the size of roots and shoots about 8.68% and 5.05% respectively as compared to the control. However, the concentration of heavy metals was increased at treatment four, the sizes of root and shoot are the same at two and three treatmens. Heavy metal organs demonstrated a different affinity to take up heavy metals. In all the plant organs, the concentrations of heavy metals increased as the metal contents in the soil increased.

Key words: Hordeum bulbosum, Heavy metals, Germination, Phytoremediation.

Introduction

Heavy metal pollution of soil is one of the most important environmental problems throughout the world. In fact, heavy metals have a significant toxicity for humans, animals, microorganisms and plants (Bodar et al., 2006). Although a small portion of heavy metals in soils is derived from natural processes (e.g. bedrock weathering), a much higher amount originates from the anthropogenic sources such as mining and smelting industry, use of mineral fertilizers and pesticides and sewage sludge application (Komárek et al., 2008). Therefore, remediation of old industrial sites is a necessary step in order to meet the demand for urban housing, office and leisure spaces. Decreasing the concentration of these potentially toxic compounds is also beneficial to the environment. Many methods including removal. incineration and removal followed by thermal desorption have been used for cleaning up the contaminated soils although all these methods are costprohibitive (Joner and Levval, 2001). To more environmentally find and economically acceptable options, scientists and engineers have recently started to generate cost-effective technologies that include the use of microorganisms, biomass and live plants in the cleaning process of polluted areas (Wasay et al., 1998; Gardea-Torresdey et al., 1996). In this way, phytoremediation that is to use green plants to decontaminate the metals out of the soil is an emerging technique with advantages of being in situ, cost effective and environmentally sustainable (Lasat, 2002).

Phytotoxicity varies according to the metal and plant species and it can occur beyond a certain threshold (Shamsi et al., 2008; Markovska et al., 2009; Mihucza et al., 2012), Therefore, for achieving maximum HMs reduction and successfully establishing vegetation stable cover. various criteria must be considered. The plants should be chosen carefully so that they provide a maximum root surface area

(Aprill and Sims, 1990). Researchers have observed that some plant species are endemic to metalliferous soils and can tolerate greater rates than usual amounts of heavy metals or other toxic compounds (Peralta et al., 2000; Blaylock and Huang, 2000). Considering the cost of more extensive testing, germination tests have frequently been used in selecting the species to be grown on the contaminated sites and most of these studies have been conducted using seedlings or adult plants (Gratton et al., 2000; Pichtel et al., 2000). In a few studies, the seeds have been exposed to the contaminants (Claire et al., 1991; Xiong, 1998). In order to explore an acceptable option for the biological control of small contaminated sites polluted by heavy metals, phytoremediation potential of Hordeum bulbosum has been considered as a local plant species in reducing the extent of contaminated soils with heavy Takestan industrial metals in citv. Therefore, the germination, growth and heavy metal uptake of H. bulbosum were investigated in the contaminated soils. Furthermore, using germination rate as an indicator of subsequent establishment and growth of plant in the contaminated soils was investigated.

Materials and Methods Soil preparation

The soil for the greenhouse experiments was obtained from Takestan industrial city, an important industrial area of Qazvin (Northern Iran). The area is influenced by the industrial waste water; then, the soil is highly polluted with heavy metals. Soil samples were collected at four different points distributed along the wastewater channel from its mouth to 1 km landward. Soil samples of treatments one, two and three were taken outside of the channel in different distances and soil sample of treatment four was taken at the mouth of channel. Soil samples were divided into 4 groups considering the metal concentration in each group with 5 pots (replicates) (Table 1).

Soil sampling was obtained from the depth of 0-40 cm and mixed (Besalatpour et al., 2008). All soil samples were sieved to 4 mm and moisture contents were adjusted to water-holding 70% capacity. Uncontaminated soil was used as the control throughout the study. Selected physical and chemical properties of contaminated and uncontaminated soil samples are listed in (Table 2). The soil texture (clay-loam) was determined by hydrometer method (Day, 1982), CEC (Cation Exchange Capacity) was measured by the method of Bower and Hatcher (1966), pH was determined in a 1:5 soil to distilled water slurry after 1 hour of using digital agitation а pH-meter (Thomas, 1996), Electrical Conductivity (EC) was measured using an EC-meter (Rhoades, 1996), total soil N was analyzed colorimetrically with a continuous flow ion analyzer following wet digestion in sulfuric acid (Bremner, 1996), organic carbon was measured by the Walkley-Black method (Nelson and Sommers, 1996). The $CaCO_3$ equivalent was determined by neutralizing with HCl and back titration with NaOH (Black et al., 1965). Heavy metal (Pb, Zn, Cu and Fe) concentrations were determined as follows: dried soil samples were passed through a 2 mm diameter sieve. About 100 mg dry sediment was digested with HNO₃ and HCl (3:1) in a microwave oven. After mineralization, the samples were diluted, filtered and then analyzed for heavy metals using ICP/OES (Model GBC, 2004).

Table 1. Geographic coordinates and distance of four treatments from the wastewater channel mouth (m)

Treatment	Distance (m)	Coordinates
1	1000	35° 41′ 13"N 48° 38′28"E
2	500	35° 34′ 27"N 48° 27′16"E
3	200	35° 33′ 61"N 48° 42′21"E
4	0	35° 25′ 33"N 48° 42′17"E

Pot experiment

Seeds were obtained commercially and about 1 kg of soil samples was added to pots (diameter 20×diameter plastic 15×height 35 cm). A paper filter was placed at the bottom of each pot to prevent the soil escaping from the drainage holes. 25 seeds of the plant were placed evenly throughout each pot at least 1 to 2 cm from the edge. The pots were watered from the top during the germination period so that the soil moisture was kept near 70% of capacity. each treatment. field For germination was monitored closely over weeks of the trials. the first two Germinated seeds were counted daily according to the method proposed by Maguire (1962) and two germination parameters were determined: (1) final germination percentage (number of germinated seeds in each pot) and (2) Germination Rate (GR, a measure of germination speed with lower values indicating faster germination). Germination rates were calculated as follows:

$$RG = \sum N_i D_i$$

where, N_i is the number of germinant daily, D_i is the number of days from the initial sowing. In each trial, the plants were grown for 6 weeks after the germination period in the greenhouse. The minimum and maximum temperature was 21 and 28C°, respectively. The seedlings were harvested at the end of growing trial. The roots were carefully washed with deionized water to get rid of the sand. The plants were separated into root and shoot and dried in an oven at 70°C for 48 h. Then, the biomass (dry weight) was determined. Dried samples were ground, digested in the concentrated HNO₃ for 14 h, 12 h at room temperature first, then 2 h at 60°C, and further digested with concentrated HNO₃-HClO₄ (3:2, v/v) for 3 h at 140-160°C. After cooling, the extract was diluted with NHCl and made up to 25 ml.

Statistical analysis

Analyses were carried out using SPSS release 16.0. Data were tested for normality with the Kolmogorov–Smirnov test and for variance homogeneity with the Levene test (P<0.05) (Mansourfar, 2006). Variations in heavy metal concentrations in the soils and plant organs were compared

by one-way analysis of variance. Tukey's t-test between means was calculated only if F-test was significant at the 0.05 level of probability. A probability of 0.05 or lower was considered as significant.

Results

Soil properties

Soils from the contaminated and uncontaminated sites had clay loam textures and electrical conductivity of soils were ranged between 2.30-2.66 dSm⁻¹. pH varied between 8.3-8.8 (Table 2). The results revealed that the soils are similar in of physical and chemical terms characteristics. Concentrations of heavy metals are listed in Table 3, the highest concentration was measured for treatment 4 and treatment 1 had the lowest heavy metal concentration.

Table 2. Physical an	l chemical	characteristics	of s	soil	samples	were	collected	for	the
greenhouse treatments									

Site	Texture	CEC	N%	OC%	EC	лU	CaCO ₃
	Texture	(meq)	1 \ %	00%	$(dS m^{-1})$	pН	(%)
1	Clay loam	36.30	0.13	0.14	2.55	8.4	12.21
2	Clay loam	35.40	0.12	0.14	2.50	8.3	12.00
3	Clay loam	36.20	0.14	0.15	2.66	8.4	12.50
4	Clay loam	35.00	0.15	0.16	2.30	8.6	13.60
Control	Clay loam	29.30	0.10	0.16	2.40	8.8	15.30

Soils were sampled from 0 to 40 cm depth with a 5 cm-diameter hand-driven corer.

Table 3. Total concentration of heavy metals (mgkg⁻¹, n=5) for contaminated treatments based on soil collected from the industrial area

Treatment	Pb	Zn	Cu	Fe
1	720.12±16.00 a	520.11±16.72 a	21.83±3.00 a	245.13±52.20 a
2	1095.00±96.11 b	1633.42±25.09 b	50.40±2.73 b	990.77±64.20 b
3	2600.42±94.70 c	2208.03±33.00 c	73.00±5.61 b	1563.9±130.1 c
4	5571.35±83.22 d	2823.17±46.11 c	115.62±77.4 c	2588.6±175.66 d

Results are mean±SE. Different letters indicate significant difference between treatments (Tukey's test).

Effects of heavy metals on seed germination

Table 4 shows the effects of various concentrations of Pb, Zn, Cu, and Fe on seed germination of *H. bulbosum* grown in the contaminated and uncontaminated soils. In general, there was a reduction in seed germination as metal concentrations in the soils increased. Significant decreases (P<0.05) of germination rate and percentage with the increase of heavy metal concentrations in the solution have been observed. These parameters at treatment one are the same with those of the control and at treatment four, the parameters are as high as 37.33% and 38.20% of the control, respectively. At treatment one. the root length is significantly different from the control decreasing to a half of the control. At treatment four (the highest content of metals), the root length is only as long as one tenth of the control.

Effects of heavy metals on root and shoot growth

Although the response to heavy metal concentration varied between root and shoot length, it demonstrated an overall dose dependent response to the contaminated soil and a decrease in the root and shoot growth was evident as compared to the control treatment. At treatment one (the lowest concentration), the reduction in root and shoot size was about 8.68% and 5.05% respectively as compared to the root and shoot size of the control. However, the concentration of heavy metals was increased at treatment four, root and shoot sizes are the same at treatments two and three.

According to the length difference of root and shoot between the lowest heavy metal concentration and control, root seems to be more sensitive to heavy metals than the shoot. The data corresponding to the biomass with various treatments are presented in (Table 4). Similar to the seed germination and root and shoot length, the mean biomass is significantly different from the treatment four decreasing to the half of the control. It is interesting to note that the response of *H. bulbosum* to the heavy metals seems to be a combination of some stimulating and inhibiting effects and the biomass reached its minimum at treatment four.

Table 4. Germination rate, germination percentage (after 14 days of treatment), biomass (DW), roots and shoots length of *H. bulbosum* at five different treatments (at the end of growing trial)

Tracture ant	Germination	Germination	Root Length	Shoot Length	Biomass	
Treatment	Rate (%)	Percentage	(mm)	(mm)	(mg)	
1	93.67±0.00 a	94.00±1.64 a	40.32±2.20 a	35.71±2.10 a	50.72±0.92 a	
2	66.55±1.70 b	62.40±1.47 b	22.37±2.50 b	25.09±2.00 b	33.41±1.36 b	
3	50.13±1.55 c	44.81±1.50 c	20.40±1.30 b	20.23±1.70 b	30.50±1.54 b	
4	37.33±1.40 c	38.20±1.74 c	18.61±1.20 b	15.12±1.22 b	30.80±1.00 b	
Control	100.00±0.00 a	100.00±1.20 a	48.00±2.16 a	39.51±1.54 a	71.84±2.09 a	

Results are mean±SE. Different letters indicate significant difference between treatments (Tukey's test).

The seedling heavy metal uptake

The heavy metal concentrations in the roots and shoot of *H. bulbosum* are listed in (Table 5). Heavy metal concentrations were significantly higher in the root tissue than aboveground components (leaf and stem) and plant organs demonstrated a different affinity to take up heavy metals. It is interesting to note that heavy metal concentration in the plant is significantly affected by the content supplied in the soil

and the plant tissue as well as the interaction between two factors (Table 5). Mean values for organs showed the maximum accumulation at treatment 4 that showed a significant accumulation (P<0.05) comparing with other treatments. However, data demonstrated an increase in the concentration of heavy metals, there was no significant difference between content of heavy metals in the root and shoot between treatments 2 and 3.

Table 5. Heavy metal concentration (mg/kg) in seedling root and shoot tissues of <i>H</i> .
<i>bulbosum</i> at the end of growing trial

Treatment	Pb		Zn		Cu		Fe	
	Root	Shoot	Root	Shoot	Root	Shoot	Root	Shoot
1	903.10	228.66	747.22	288.30	79.00	66.40	330.56	127.00
	±14.00	±7.09	±10.05	±9.53	±5.40	±7.33	±14.00	± 4.76
	C-b	C-a	C-b	C-a	C-a	C-b	D-a	C-b
2	4321.00	978.20	2208.46	823.07	244.39	103.57	1240.22	1005.00
	±53.33	±11.35	±18.00	±7.77	±8.90	±8.00	±17.30	±14.23
	B-b	B-a	B-b	B-a	B-a	B-b	C-a	B-b
3	4799.15	1054.06	2401.52	1444.82	253.28	124.04	2247.27	1340.46
	±47.18	±13.40	±24.70	±16.24	±10.72	±6.42	±24.00	±23.08
	B-b	B-a	B-b	B-a	B-a	B-b	B-a	B-b
4	6420.50	2300.92	4900.10	2805.38	724.06	446.50	4358.12	1970.10
	±76.00	±25.00	±57.05	±29.50	±11.50	±12.37	±63.90	±33.50
	A-b	A-a	A-b	A-a	A-a	A-b	A-a	A-b
Control	ND	ND	ND	ND	ND	ND	ND	ND

Results are mean±SE. Different capital letters in each column indicate significant differences between treatments. Different lower case letters in each row indicate significant differences between organs (Tukey's test). ND= NOT Detected/ Below detectable range.

Discussion and Conclusion

In the current research study, the effect of heavy metals on the seed germination and growth of *H. bulbosum* was assessed. In general, there was a reduction in seed germination and growth as metal concentration in the soils increased. Claire *et al.*, (1991) obtained similar results in a study using nickel and other heavy metals on cabbage, lettuce, millet, radish, turnip and wheat. Similar results have also been observed in *Pisum sativum* L. (Sabrine *et al.*, 2010), *Sinapis alba* (Fargasova, 1994), *Brassica pekinensis* (Xiong, 1998), lettuce and radish (Nwosu *et al.*, 1995) and *Medicago sativa* (Peralta *et al.*, 2000). Decrease in the root and shoot growth was evident as compared to the control treatment. Arduini et al., (1994) found the similar result in a research on morphology of Pinus pinea and Pinus pinaster using cadmium and copper. Peralta et al., (2004) reported that a reduction in chlorophyll could diminish the aboveground organ growth. In current study, the data revealed that the root seems to be more sensitive to heavy metals than the shoot because an increase in heavy metal concentrations of soil leads to a root elongation reduction more than the shoot. One of the explanations for the roots to be more responsive to the contaminants in environment might be that the roots were the specialized absorptive parts so that they were affected earlier and subjected to accumulation of more toxic elements than any other parts. This could also be the main reason for that the root length was usually used as a measure for determining heavy metal-tolerant ability of plants. Xiong (1998) found a similar effect using lead in B. pekinensis seed.

The highest total dry biomass (roots plus shoots) was observed in the control treatment followed by treatment one and the lowest total dry biomass was seen in treatment four (the highest heavy metal concentration). Decreasing the dry biomass might be due to a decrease in water absorption in plant tissues by toxic metals causing undesirable impacts on the plant growth (Fuentes *et al.*, 2006). Similar results have also been reported in the studies of Inckot *et al.*, (2011), Papazoglou *et al.*, (2005) and Peralta *et al.*, (2004).

Regarding the concentration of metals in the aerial tissues, root played a more important role and the absorption of metals in the root was higher. Therefore, the metals investigated in the root of plant showed a significant increase and a relationship between the plant root and metal accumulation was clearly visible. So, the roots of plant could be considered as an index for Pb, Zn, Cu and Fe absorption. The relatively low accumulation of heavy

metals in the aboveground tissues was probably due to the need of plants to prevent toxicity to the photosynthetic apparatus as suggested by other authors (Stoltz and Greger, 2002; Bragato et al., 2006). In all the tissues (root and shoot), concentrations of heavy metals the increased without exception as the metal contents in the soil was increased. Published data revealed that heavy metal content in the plant is a function of heavy metal concentration in the soil. Kabata-Pendias and Pendias (1984) reported that the plants would take up the metal via the root system when the soil is contaminated with heavy metals. In this study, root and shoot are directly subjected to heavy Thus. the heavy metals. metal concentration in the tissues is more closely related to that of the soil.

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